

Synthesis of methyl 3-amino-2-hydroxy-4-phenylbutanoates, important core intermediates for peptide mimics possessing biological activities

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Abstract—Core intermediates for the synthesis of anti-cancer agents, inhibitors of aminopeptidases and HIV proteases, 3-amino-2-hydroxy-4-phenylbutanoates were synthesized by using H-C(CN)₂O-SiR₃ as a key reagent. © 2001 Elsevier Science Ltd. All rights reserved.

An unusual amino acid **1a**, (2*S*,3*R*)-*N*,*O*-protected-3-amino-2-hydroxy-4-phenylbutanoate as well as its 2*R*-diastereomer **1b**, is one of the important core intermediates¹ of several naturally occurring or artificially synthesized anti-cancer agents, inhibitors of aminopeptidases and HIV proteases² (Scheme 1). To synthesize **1** or the related compounds, several methods including the nucleophilic addition reaction to the *N*-protected-2-amino-3-phenylpropionaldehydes **2** by a masked formyl anion **3**³ have been reported. Although the reaction is one of the most straightforward synthetic routes, the methods generally consisted of *several steps* including an unmasking procedure under vigorous conditions.

We report the synthesis of the methyl (2S,3R)-N,O-

protected-3-amino-2-hydroxy-4-phenylbutanates **9a–11a** as shown in Scheme 2 by using 2-(*tert*-butyldimethylsiloxy)malononitrile (H-MAC-TBS,⁴ **4m**) or 2-(triisopropylsiloxy)malononitrile (H-MAC-TIPS, **4n**) *in a one-portion manipulation* directly from **2Z** or **2B**.⁵

We first attempted the reaction of **2Z** with H-MAC-MOM (**4I**) in acetonitrile with various catalysts or promoters in the presence of trapping reagents such as acetic anhydride or diketene. However, no *O*-acylated derivatives of the adducts **6** (R¹=Cbz, R²=MOM) were obtained, and more than 50% of the starting material **2Z** was recovered probably because of a reversible reaction (Scheme 3). Therefore, we examined **4m** instead of **4l** since a silyl group could be migrated to generate **7** or **8**, which inhibits the reversible reaction.

R¹ NH
$$Z^1$$
 Z^2 $Z^$

Scheme 1.

Keywords: amino acids and derivatives; amino aldehydes; acyl cyanides; acyl anion equivalents; biologically active compounds.

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We first carried out the *one-pot but stepwise manipulation* as follows. Anticipating the generation of 7 or 8, a mixture of DL-2Z and 4m in acetonitrile was stirred for 10 minutes, then pyridine and methanol were added. However, the *stepwise manipulation* under various conditions did not give the corresponding methyl esters DL-9.

Next, we examined the *one-portion manipulation*. In methanol in the presence of pyridine, the desired methyl esters **9** were obtained directly from **2Z** in 36% yield (entry 1 in Table 1). Encouraged by this result, further various conditions were examined. When 4-(*N*,*N*-dimethylamino)pyridine (DMAP) was used as base

$$\begin{array}{c} \textbf{2} + \textbf{H} & \textbf{CN} \\ \textbf{CN} & \textbf{In One-portion} \\ \textbf{CN} & \textbf{4} \\ \\ \textbf{2Z: } R^1 = \text{benzyloxycarbonyl (Cbz)} \\ \textbf{2B: } R^1 = \textit{tert-butyloxycarbonyl (Boc)} \\ \textbf{4I: } R^2 = -\text{CH}_2\text{OCH}_3 \text{ (MOM)} \\ \textbf{4m: } R^2 = -\text{Si}^{\text{i}}\text{Pr}_3 \text{ (TIPS)} \\ \end{array} \begin{array}{c} \textbf{9a: } R^1 = \text{Cbz, } R^2 = \text{TBS} \\ \textbf{10a: } R^1 = \text{Cbz, } R^2 = \text{TBS} \\ \textbf{11a: } R^1 = \text{Cbz, } R^2 = \text{TIPS} \\ \textbf{12a: } R^1 = \text{Cbz, } R^2 = \text{TIPS} \\ \textbf{12a: } R^1 = \text{Cbz, } R^2 = \text{H} \\ \textbf{12b: } R^1 = \text{Cbz, } R^2 = \text{H} \\ \textbf{12b: } R^1 = \text{Cbz, } R^2 = \text{H} \\ \end{array}$$

Scheme 2.

Table 1. Reaction of (\pm) -2Z and 4m

Entry	Solvent	Base	Temp. (°C)	Period (h)	9		13
					Yield	9a:9b	Yield
1	МеОН	Pyridine	rt	2	36	72:28	63
2	MeOH	DMAP	-25	12	45	63:37	51
3	Et ₂ O	Pyridine	rt	48	Trace	_	_
4	Et ₂ O	Imidazole	rt	2	67	78:22	20
5	Et_2O	DMAP	0	5	83	79:21	Trace
6	Et ₂ O	PPY	rt	2	80	78:22	Trace
7	Et_2^2O	PPY	0	5	868	79:21	Trace
8	Et ₂ O	PPY	-25	12	65	79:21	Trace

A representative procedure (entry 7): A mixture of DL-2Z (23.4 mg, 0.083 mmol), 4m (32.4 mg, 0.165 mmol), PPY (12.2 mg, 0.083 mmol) and methanol (10 µl, 0.247 mmol) in ether (2 ml) was stirred at 0°C for 5 h. The resulting mixture was concentrated in vacuo, and the residue was purified by column chromatography on silica gel (hexane/ethyl acetate=4:1) to afford 9a and 9b (total 32.4 mg, 0.071 mmol, 86% yield).

Scheme 3.

instead of pyridine, the yield was slightly increased (entry 2). Since the cyanohydrin DL-13 was obtained in briefly 50–60% yield as a major byproduct in entries 1 and 2, we considered that the formation of 13 could be inhibited in less polar solvent than in methanol. Therefore, we used ether⁷ as a solvent with 3 equivalents of methanol with various bases. Using 4-pyrrolidinopyridine (PPY) gave the best result (entry 6) of all the four examinations (entries 3, 4, 5, and 6). Next, we carried out the reactions with PPY at room temperature (rt), 0°C and -25°C (entries 6, 7 and 8, respectively). As shown in entry 7, DL-9 was obtained at 0°C in satisfactory yield. The distereomeric ratio of DL-9a and DL-9b was 79:21.8,9 We also carried out the reaction of the optically active compound D-2Z under the same conditions as entry 7. No epimerization of the C_3 position of either 9a or 9b occurred in this experiment.¹⁰

The syntheses of **10** and **11** having alternative protecting groups were also examined under similar conditions to entry 7. Both were obtained in 80 and 85% yields, respectively, with similar diastereoselectivity (**10a:10b** = **11a:11b** = 79:21). Stereochemistry of the compounds **9–11** was determined by well-known protection/deprotection procedures to convert them to the known compounds **12a** and **12b**. ^{1,11}

In conclusion, we have synthesized methyl *N*-protected-3-amino-2-siloxy-4-phenylbutanoates including optically active ones in a one-portion manipulation in high yields with good diastereoselectivity. Further synthetic studies for biologically active compounds are now in progress.

References

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- 5. The aldehydes DL-2Z, D-2Z, and DL-2B were prepared according to the available procedure. Fehrentz, J.-A.; Castro, B. *Synthesis* 1983, 676–678.

- 6. The conditions we first examined were referred from our previous publications. (a) Nemoto, H.; Kubota, Y.; Yamamoto, Y. *J. Chem. Soc., Chem. Commun.* **1994**, 1665–1666. (b) Nemoto, H.; Ma, R.; Ibaragi, T.; Suzuki, I.; Shibuya, M. *Tetrahedron* **2000**, *41*, 1463–1468.
- 7. The reaction rates of both MAC anion and cyanide anion to 2Z could be strongly affected by the polarity of solvents. In fact, we observed that the aldehyde 2Z was consumed within a short period in acetonitrile as well as in methanol. In hexane or toluene as a representative non-polar solvent, more than 10 hours were required to finish the reaction. However, the ratio of the desired compound 9a to the cyanohydrin 13 was independent from the polarity of solvent. We chose ether since the ratio in ether was higher than the ratios in other solvents.
- According to the following publications, the major isomer 9a is the product by non-chelation control. (a) Herrantz, R.; Castro-Pichel, J.; Vinuesa, S.; Garcia-Lopez, M. T. J. Org. Chem. 1990, 55, 2232–2234. (b) Reetz, M. T.; Drewes, M. W.; Harms, K.; Reif, W. Tetrahedron Lett. 1988, 29, 3295–3598.
- 9. We also carried out the reaction based on entry 7 in the presence of several Lewis acids such as zinc chloride, magnesium bromide, triisobutylalminium and tin(II) triflate. However, the diastereoselectivity of 9a was not optimized. Furthermore, chemical yields were dramatically decreased in some cases.
- 10. No epimerization was confirmed by HPLC analysis of the desilylated derivatives (+)-12a and (+)-12b (Daicel OD chiral column 4.6 mm $\Phi \times 250$ mm length, hexane/ethanol=15:1, flow rate=0.7 ml/min. (+)-12a: rt=24.7 min, (-)-12a: rt=30.6 min, (+)-12b: rt=34.3 min, (-)-12b: rt=31.9 min).
- 11. To determine the relative stereochemistry of all the compounds, we carried out the following reactions.

9a or $9b \rightarrow 12a$ or 12b (Bu₄NF in THF at 0°C for 30 min, 88% yield)

9a or $9b\rightarrow 10a$ or 10b (Pd(OH)₂/H₂, Boc₂O, in MeOH at rt for 2 h, 91% yield)

11a or 11b \rightarrow 12a or 12b (Bu₄NF in THF at 0°C for 30 min, 91% yield)

9a: colorless oil; FT-IR (CHCl₃): 3437, 2953, 1753, 1717, 1504, 1146, 840 cm⁻¹; ¹H NMR (300 MHz): δ 7.40–7.10 (m, 10H), 5.17 (brd, J=8.5 Hz, NH), 5.01 (d, J=12.3 Hz, 1H), 4.97 (d, J=12.3 Hz, 1H), 4.41–4.29 (m, 1H), 4.23 (d, J=1.4 Hz, 1H), 3.64 (s, 3H), 2.87 (d, J=7.5 Hz, 2H), 0.96 (s, 9H), 0.10 (s, 3H), 0.05 (s, 3H). ¹³C NMR (75 MHz): δ 172.2, 155.7, 137.5, 136.5, 129.2, 128.5, 128.5, 128.1, 128.0, 126.6, 72.1, 66.6, 55.5, 52.0, 38.1, 25.8, 18.4, -4.7, -5.3; EI-HRMS calcd for $C_{25}H_{35}NO_5Si$ (M⁺): 457.2285. Found 457.2302.

9b: colorless oil; FT-IR (CHCl₃): 3443, 2954, 1753, 1719, 1507, 1254, 1150, 838 cm⁻¹; ¹H NMR (300 MHz): δ 7.39–7.10 (m, 10H), 5.03 (s, 2H), 4.86 (brd, J=7.6 Hz, NH), 4.45 (d, J=3.1 Hz, 1H), 4.50–4.25 (m, 1H), 3.64 (s, 3H), 2.87–2.62 (m, 2H), 0.93 (s, 9H), 0.08 (s, 3H), 0.01 (s, 3H). ¹³C NMR (75 MHz): δ 171.7, 155.6, 137.3, 136.4, 129.4, 128.5, 128.4, 128.1, 128.0, 126.7, 73.3, 66.7, 55.0, 51.8, 35.5, 25.7, 18.3, –5.0, –5.5; EI-HRMS calcd for $C_{25}H_{35}NO_5Si$ (M*): 457.2285. Found 457.2249.